# HUMULENES AND OTHER CONSTITUENTS OF FERULA LATIPINNA

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ABSTRACT.—Chemical analysis of *Ferula latipinna* yielded five new sesquiterpenes, carotanes 1 and 6 and the humulenes 12, 13, and 16 as well as three new phenylpropanes 20, 22, and 24, a phenolic derivative 25, scopoletin, isoscopoletin, diosmetin, veratric, and *p*-methoxybenzoic acids, and a new trimethoxyphenanthrene 27. The structures of the new compounds were elucidated by spectroscopic methods and chemical reactions.

Some 130 species of *Ferula* L. (Umbelliferae) are to be found in the Old World (1), from the Mediterranean area to Central Asia and also in Macronesia. In the Canary Islands three species have been reported to date: *Ferula linkii* Webb, *Ferula lancerottensis* Parl, and *Ferula latipinna* Santos (2). The latter species is endemic to La Palma though it may also occur in La Gomera. It is clearly differentiated from the other two species, most distinctively by its broad pinnules. It grows on steep mountain sides and in clearings in the evergreen forest zone on the north side of the island where it is prized by the local goatherds for its galactophorous properties.

Ferula species are rich in sesquiterpenes (3), including  $\alpha$ -humulenes from Ferula juniperina Eug. Kor. (4–6), Ferula xeromorpha Eug. Kor. (7,8), and Ferula tschatcalensis M. Pimen. (9),  $\gamma$ -humulenes from Ferula ceratophylla Regl. ex Schmalh. (10), apiene esters from Ferula haussknechtii Wolff ex Rech. (11), and himachalenes from F. xeromorpha (12). Similar substances have now been isolated from F. latipinna.

## **RESULTS AND DISCUSSION**

Stilbenes and phenylpropanes from *F. latipinna* were reported earlier (13). The current study of the EtOH extract of the roots and aerial part of this plant species afforded sesquiterpenes **1–11**, eight of which had previously been found in other *Ferula* species:  $10\beta$ -hydroxy- $5\alpha$ -*p*-hydroxybenzoyloxy- $1\alpha$ -angeloyloxydauc-2-ene [3],  $10\beta$ -hydroxy- $5\alpha$ -*p*-anisoyloxy- $1\alpha$ -angeloyloxydauc-2-ene [4] (14), lapidin [5] (15), lapiferin [7], lapiferol [8] (16), felikiol 3-angelate [9], and webiol 3-angelate [11] (17). Felikiol [10] was found for the first time as a natural product. Compounds 1 and 6 have not hitherto been reported. Three new humulenes 12, 13, and 16, himachalol [17] (18), five phenylpropanes including myristicin [18], crocatone [19], and compounds 20, 22, and 24, a new phenolic derivative 25, a new trimethoxyphenanthrene 27, and the coumarins scopoletin and isoscopoletin, the flavonoid diosmetin, and veratric and *p*-methoxybenzoic acids were also obtained. These are the first coumarins and flavonoids to be found in the Canary Islands *Ferula* species. The known substances were identified by comparison with authentic samples.

Compound 1, mp 77–78°. Ms did not directly establish the molecular formula; however, the fragment at m/z 275  $[M - C_3H_7 - C_5H_8O_2]^+$  corresponds to the formula  $C_{25}H_{38}O_5$ , with the loss of an isopropyl radical and angelic acid. The ease with which the isopropyl radical was lost indicates that compound 1 must be a carotaneskeleton sesquiterpene with the hydroxy at C-10. The ir clearly shows the presence of hydroxy groups (3580, 3460 cm<sup>-1</sup>), esters (1700, 1260, 1230 cm<sup>-1</sup>), and double bonds (1640, 840 cm<sup>-1</sup>). The <sup>1</sup>H nmr is similar to that of tingitanol (19), differing in that 1 does not have a signal for a geminal hydroxy proton at C-8.

0

Ŗ1

Ř2  $R_1 = OAc, R_2 = OAng$ 

ËΝ

8  $R_1 = R_2 = OH$ 

HO

7



- $R_1 = R_2 = OAng$ 1
- 3  $R_1 = OAng, R_2 = 0-p-hydro$  $4 <math>R_1 = OAng, R_2 = p-anisate$  $R_1 = OAng, R_2 = 0-p-hydroxybenzoate$
- 5  $R_1 = O, R_2 = OAng$
- **6**  $R_1 = O, R_2 = 0$ -vanillate



R=OH

10





12 R=veratrate





13  $R_1$  = veratrate **14**  $R_1 = H$ **15**  $R_1 = p - Br - C_6 H_4 - CO$ 



16 R=veratrate





18



19  $R_1 = O, R_2 = OMe$ **20**  $R_1 = R_2 = OH$  $R_1 = R_2 = OAc$  $R_1 = OH, R_2 = OMe$ 21 22 **23**  $R_1 = OAc, R_2 = OMe$ 







The ester group positions were ascertained by the following reactions (Scheme 1). Treatment of **1** with MCPBA yielded the epoxyderivative **2**. The <sup>1</sup>H-nmr spectrum showed signals for an angular methyl at  $\delta$  1.26 (3H, s, H-14), two tertiary methyls at  $\delta$  0.88 (3H, s, H-12) and 0.91 (3H, s, H-13), an oxirane methyl at  $\delta$  1.43 (3H, s, H-15) with the corresponding proton at  $\delta$  3.00 (1H, d, J = 5.4 Hz, H-2), and signals for the epoxyangeloyloxy moiety. Alkaline hydrolysis of **2** gave a triol with physical and spectral data (ir, <sup>1</sup>H nmr) identical to those of lapiferol [**8**] (16). Compound **1** was given the name 8-desoxytingitanol.



Compound 6 is a lapidol derivative. Aromatic acyl group absorptions were observed in its ir spectrum, and <sup>1</sup>H, <sup>13</sup>C nmr, and ms confirmed the presence of a vanillate group. Alkaline hydrolysis of 6 gave lapidol (15) and vanillic acid.

Sesquiterpene 12 was isolated as a colorless oil, composition  $C_{24}H_{34}O_4$ ; ir showed absorptions (1690, 1600, 1590, 1510 cm<sup>-1</sup>) for an aromatic acyl group which was identified as a veratrate from its <sup>1</sup>H-, <sup>13</sup>C-nmr, and ms data. The <sup>1</sup>H nmr of 12 showed two vinyl protons and vinyl methyls indicating the presence of two double bonds. Its composition, the two double bonds, and the nature of the methyl groups suggest that this product probably has a monocyclic humulene skeleton (20–22). The position of the aromatic acyl group at C-6 was confirmed by the <sup>1</sup>H-nmr spectrum where the doublet at  $\delta$  5.50 (1H) was coupled with the signal of H-5 at  $\delta$  5.40. NOe experiments on 12 showed that the  $\Delta^{1,10}$  and  $\Delta^4$  double bonds have *E* and *Z* configurations, respectively, and that the stereochemistry of the acyl group at C-6 is  $\beta$  (11). NOe interactions were observed for H-1 and H-6, H-7 $\alpha$  and H-3 $\alpha$ , for H-5 and H-12 and H-15, for H-6 and H-3 $\alpha$ , H-7 $\alpha$  and H-13, for H-14 and H-2 $\beta$ , and for H-15 and H-5.

Compounds 13 and 16 have <sup>1</sup>H- and <sup>13</sup>C-nmr spectra analogous to those of 12 (see Table 1), differing in that the former have a 1(10)-oxirane ring. The disposition of this ring was determined by spectroscopic analysis. Alkaline hydrolysis of 13 gave veratric acid and a monodesacyl derivative 14 with the empirical formula  $C_{15}H_{26}O_2$ , analogous

to that described first by Itokawa and co-workers (23, 24) and later by Miski *et al.* (11). Treatment of **13** with Cl-*p*-bromobenzoyl gave *p*-bromobenzoyl ester **15** identical to that described by the above authors.

Compound **16** proved to be analogous to **13**, the only difference being the epoxide stereochemistry. The structure was confirmed by  $^{13}$ C nmr [62.4 (C-1) and 62.9 (C-10) compared with 60.8 for both carbons in compound **13**], and comparison with known compounds (24).

The phenylpropanoids **20** and **22** were purified as acetyl derivatives. Compound **20** gave a diacetate **21**, molecular formula  $C_{14}H_{16}O_6$  (m/z 280 [M]<sup>+</sup>). In the <sup>1</sup>H-nmr spectrum signals for two acetyl groups were observed at  $\delta$  2.03 (3H) and 2.27 (3H). Compound **22** gave a monoacetate **23**, molecular formula  $C_{13}H_{16}O_5$  (m/z 252 [M]<sup>+</sup>). The <sup>1</sup>H nmr exhibited signals for an acetyl and a methoxy group at  $\delta$  2.06 (3H) and 3.89 (3H), respectively. Hydrolysis of **21** followed by treatment with  $CH_2N_2$  gave **22** while hydrolysis of **23** gave an alcohol identical (ir, <sup>1</sup>H nmr, ms) to that obtained by NaBH<sub>4</sub> reduction of **19**, earlier isolated from *Ferula ugamica* G. Kor. (25).

Phenylpropanoid **24** was a crystalline substance,  $mp 95-96^\circ$ ,  $C_{11}H_{12}O_5$  (m/z 224 [M]<sup>+</sup>), which turned dark blue when treated with FeCl<sub>3</sub>. Absorption bands for carbonyl and aromatic groups were visible in the ir spectrum (1650, 1605, 1504 cm<sup>-1</sup>), and <sup>1</sup>H nmr clearly showed the presence of a methylenedioxy group and a methoxy group on an aromatic ring [ $\delta 6.10(2H)$ , 3.88(3H)]. The structure of **24** was confirmed by nOe difference spectroscopy.

The phenolic compound **25** was a colorless oil with molecular formula  $C_{10}H_{14}O_3$ (m/z 182 [M]<sup>+</sup>). In the <sup>1</sup>H-nmr spectrum, two doublets appeared in the aromatic region at  $\delta$  7.15 (2H, J = 8 Hz) and 6.79 (2H, J = 8 Hz) corresponding to protons in the ortho position on a substituted aromatic ring. A quartet at  $\delta$  3.42 (2H, J = 7 Hz) and a triplet at  $\delta$  1.84 (3H, J = 7 Hz) were assigned to an ethoxy group. Compound **25** was treated with Ac<sub>2</sub>O/pyridine to give the diacetate **26** as an oil:  $C_{14}H_{18}O_5$ ; no [M]<sup>+</sup>, m/z 221 [M -  $C_2H_5O$ ]<sup>+</sup>. The <sup>1</sup>H-nmr spectrum showed two singlets at  $\delta$  1.97 (3H) and 2.20 (3H), attributed to the protons of aliphatic and aromatic acetyls, respectively, a triplet at  $\delta$  4.44 (1H, J = 6 Hz), an acetate group geminal proton, and a doublet at  $\delta$ 4.06 (2H, J = 6 Hz) assignable to a -CH<sub>2</sub>O-. The proposed structure was confirmed by <sup>13</sup>C-nmr analysis. Compound **25** is probably an artifact of the extraction process.

The new substituted phenanthrene **27** had the molecular formula  $C_{17}H_{16}O_3$  (m/z 268 [M]<sup>+</sup>). In the <sup>1</sup>H-nmr spectrum there were signals for methoxy groups at  $\delta$  3.95, 3.99, and 4.10 (each 3H, s) and for seven aromatic protons at  $\delta$  6.75 and 6.89 (each 1H, d, J = 2.5 Hz, H-1 and H-3 or vice versa), 7.20 and 7.50 (each 1H, d, J = 8.7 Hz, H-9 and H-10 or vice versa), 7.17 (1H, dd, J = 8.7, 2.5 Hz, H-7), 7.75 (1H, d, J = 8.7 Hz, H-8), and 9.07 (1H, d, J = 2.5 Hz, H-5). The methoxy groups had to be positioned at C-2, C-4, and C-6, by comparison of the <sup>1</sup>H-nmr spectrum of **27** with those of 2,7-dihydroxy-4,6-dimethoxyphenanthrene (26).

### EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Melting points are uncorrected. Ir spectra were taken in CHCl<sub>3</sub> and <sup>1</sup>H nmr in CDCl<sub>3</sub> at 200 MHz. <sup>13</sup>C nmr were measured in CDCl<sub>3</sub> at 22.6 MHz and ms were obtained using a direct inlet system at 70 eV.

PLANT MATERIAL.—The roots and aerial part of *F. latipinna* were collected in the Gallego area, La Palma, in April 1986. A voucher specimen is deposited in the Herbarium of the Faculty of Pharmacy, Universidad de La Laguna.

ISOLATION OF COMPOUNDS.—The dried and powdered roots and aerial part of the plant (10 kg) were extracted with EtOH in a Soxhlet. The EtOH extract was concentrated in vacuo, yielding 50 g crude viscous oil. This oil was chromatographed on a Si gel column packed in  $C_6H_6$ , then eluted with  $C_6H_6/$  EtOAc mixtures of increasing polarity. The fractions obtained with an 8:2 mixture gave 1 (1.5 g), 3 (1.2

g), 4 (10 mg), 5 (800 mg), 7 (920 mg), 9 (100 mg), 11 (50 mg), 12 (960 mg), 13 (70 mg), 16 (30 mg), 17 (950 mg), 18 (10 mg), 19 (1.2 g), 20 (isolated as diacetate, 12 mg), 22 (isolated as acetate, 15 mg), 24 (22 mg), and 27 (11 mg); a 3:2 mixture gave 6 (200 mg), 8 (20 mg), 10 (25 mg), 25 (9 mg), scopoletin (20 mg), isoscopoletin (8 mg), diosmetin (1.5 g), veratric acid (2 g), and p-methoxybenzoic acid (920 mg). Sephadex LH-20 columns packed in hexane-CHCl<sub>3</sub>-MeOH (2:1:1) and/or preparative tlc (Si gel in thicknesses ranging from 1 to 5 mm) developed with hexane-EtOAc (4:1, 7:3, and 3:2) were used in the further purification of the compounds.

Compound 1.—Crystallized with di-isopropyl ether as plates; mp 77–78°; ir  $\nu$  max 3580, 3460, 1700, 1640, 1450, 1380, 1155, 1100, 1080, 1035, 975, 950, 910, 880, 840 cm<sup>-1</sup>; <sup>1</sup>H nmr  $\delta$  6.03 (2H, m, angelic acid), 5.70 (1H, d, J = 7.4 Hz, H-2), 5.23 (1H, td, J = 3, 11 Hz, H-5), 4.85 (1H, d, J = 7.4 Hz, H-1), 2.62 (1H, d, J = 11 Hz, H-6), 1.98–1.79 (4 × Me, m, angelic acid), 1.75 (3H, s, H-15), 1.13 (3H, s, H-14), 0.84 (6H, d, J = 6.7 Hz, H-12 and H-13); ms m/z (rel. int.) [M – C<sub>3</sub>H<sub>7</sub> – angelic acid]<sup>+</sup> 275 (3), 235 (4), 218 (5), 203 (4), 175 (72), 157 (14).

Epoxidation of 1.—Compound 1 (200 mg) in CHCl<sub>3</sub> (3 ml) was added to a solution of MCPBA (300 mg) in CHCl<sub>3</sub> (4 ml). The mixture was left at room temperature for 2 days and worked up in the usual way, giving the epoxyderivative 2 (150 mg): <sup>1</sup>H nmr,  $\delta$  5.17 (1H, t, J = 11 Hz, H-5), 4.97 (1H, d, J = 5.4 Hz, H-1), 3.00 (1H, d, J = 5.4 Hz, H-2), 1.56–1.28 (4 × Me, m, epoxyangelic acid), 1.43 (3H, s, H-15), 1.23 (3H, s, H-14), 0.90 (6H, d, J = 7 Hz, H-12 and H-13).

Hydrolysis of 2.—Compound 2 (90 mg) was treated with 3% NaOH in MeOH (10 ml) at room temperature, and after 4 h the mixture was worked up in the usual way, giving 8 (20 mg) (16).

Compound 6.—Mp 176–178° (hexane/EtOAc);  $[M-C_3H_7]^+$  at 359.1510,  $C_{20}H_{23}O_6$  requires 359.1494; ir  $\nu$  max 3510, 3000, 2955, 1685, 1640, 1600, 1590, 1512, 1460, 1420, 1355, 1275, 1100, 1030, 930 cm<sup>-1</sup>; <sup>1</sup>H nmr & 7.58 (1H, dd, J = 1.8, 8.2 Hz, H'-7), 7.52 (1H, d, J = 1.8 Hz, H'-3), 6.95 (1H, d, J = 8.2 Hz, H'-6), 6.01 (1H, br s, H-2), 5.83 (1H, m, H-5), 3.02 (1H, dd, J = 5, 16 Hz, H-4 $\alpha$ ), 2.57 (1H, d, J = 9.6 Hz, H-6), 2.46 (1H, dd, J = 2.5, 16 Hz, H-4 $\beta$ ), 2.00 (3H, br s, H-15), 1.39 (3H, s, H-14), 0.86 (6H, d, J = 6.7 Hz, H-12 and H-13); <sup>13</sup>C nmr & 208.3 (C-1), 166.3 (C'-1), 150.8 (C-3)<sup>1</sup>, 150.5 (C'-4)<sup>1</sup>, 146.6 (C'-5), 128.6 (C-2), 124.2 (C'-7), 121.8 (C'-2), 114.5 (C'-3), 112.0 (C'-6), 84.7 (C-10), 70.8 (C-5), 56.1 (OMe), 55.3 (C-7), 50.3 (C-6), 38.3 (C-4), 36.9 (C-9), 36.4 (C-11), 31.2 (C-8), 28.3 (C-15), 18.4 (C-14), 18.2 (C-12), 17.4 (C-13). The number of protons directly attached to each carbon was verified with the DEPT pulse sequence. Ms m/z (rel. int.)  $[M - C_3H_7]^+$  359 (7), 235 (26), 217 (9), 191 (23), 168 (36), 163 (10).

*Hydrolysis of* **6**.—Compound **6** (100 mg), dissolved in MeOH (2 ml), was treated with a 3% solution of KOH in MeOH (3 ml), at room temperature for 10 h. Usual workup afforded lapidol (6 mg) and vanillic acid (12 mg).

Compound **12**.—[**M**]<sup>+</sup> at m/z 386.2434, C<sub>24</sub>H<sub>34</sub>O<sub>4</sub> requires 386.2435; ir  $\nu$  max 3080, 3015, 2965, 2930, 1690, 1650, 1600, 1585, 1510, 1460, 1410, 1380, 1360, 1175, 1140, 1130, 1105, 1025, 930 cm<sup>-1</sup>; <sup>1</sup>H nmr  $\delta$  7.61 (1H, dd, J = 1.95, 8.5 Hz, H'-7), 7.51 (1H, d, J = 1.95 Hz, H'-3), 6.82 (1H, d, J = 8.5 Hz, H'-6), 5.47 (1H, d, J = 10.4 Hz, H-6), 5.37 (1H, br d, J = 10.4 Hz, H-5), 5.22 (1H, br t, J = 8 Hz, H-1), 3.87 (6H, s, 2-OMe), 2.80 (1H, m, H-3), 2.17 (2H, m, H-2), 1.86 (1H, dt, J = 3.2, 13.2 Hz, H-3), 1.73 (3H, br s, H-15), 1.64 (3H, br s, H-14), 1.06 (3H, s, H-12), 0.84 (3H, s, H-13); ms m/z (rel. int.) [**M**]<sup>+</sup> 386 (3), 290 (1), 206 (2), 204 (37), 189 (20), 182 (100), 167 (23); <sup>13</sup>C nmr see Table 1.

Compound **13**.—Mp 147–148° (hexane);  $[M]^+$  at m/z 402.2397,  $C_{24}H_{34}O_5$  requires 402.2404; ir  $\nu$  max 3200, 2965, 2938, 1700, 1600, 1590, 1510, 1460, 1380, 1360, 1265, 1175, 1130, 1105, 1020, 950 cm<sup>-1</sup>; <sup>1</sup>H nmr  $\delta$  7.66 (1H, dd, J = 1.5, 8.3 Hz, H'-7), 7.55 (1H, d, J = 1.5 Hz, H'-3), 6.88 (1H, d, J = 8.3 Hz, H'-6), 5.73 (1H, d, J = 10.5 Hz, H-6), 5.39 (1H, br d, J = 10.5 Hz, H-5), 3.92 (6H, s, 2-OMe), 2.84 (2H, m, H-1 and H-3), 1.82 (3H, br s, H-15), 1.39 (3H, s, H-14), 1.06 (3H, s, H-12), 0.96 (1H, s, H-13); ms m/z (rel. int.)  $[M]^+$  402 (0.86),  $[M - C_9H_{10}O_4]^+$  220 (3), 205 (2), 202 (1), 187 (2), 183 (10), 182 (100), 165 (82); <sup>13</sup>C nmr see Table 1.

Hydrolysis of 13. —Compound 13 (60 mg) dissolved in MeOH (3 ml) was treated with a 3% solution of KOH in MeOH (4 ml) at room temperature for 24 h. Usual workup and subsequent purification by cc on Si gel column using  $C_6H_6$ -MeCN (9:1) as eluent yielded alcohol 14 (20 mg): <sup>1</sup>H nmr,  $\delta$  5.37 (1H, br d, J = 10.2 Hz, H-5), 4.28 (1H, d, J = 10.2 Hz, H-6), 2.84 (1H, dd, J = 3.4, 10.7 Hz, H-1), 1.82 (3H, br s, H-15), 1.30 (3H, s, H-14), 0.96 (3H, s, H-12), 0.86 (3H, s, H-13).

<sup>&</sup>lt;sup>1</sup>Assignments interchangeable.

Carbon <sup>a</sup>		Compound		
		12	13	16
1		124.6	60.8	62.4
2		25.1	29.3	28.7
3		32.8	25.1	25.1
4		142.3	141.4	141.8
5		122.2	122.6	122.7
6		74.5	74.0	75.2
7		36.6	37.5	37.3
8		23.6	19.8	22.2
9		35.8	38.7	37.3
10		136.0	60.8	62.9
11		37.7	37.6	37.3
12	]	24.6	24.5	24.2
13		24.0	23.0	23.1
14		18.8	16.7	21.8
15		23.0	23.5	23.2

TABLE 1.  $^{13}$ C-nmr Data for Compounds 12, 13, and 16.

<sup>a</sup>Veratrate: 166.0 (C'-1), 123.5 (C'-2), 111.2 (C'-3), 148.7 (C'-4), 152.9 (C'-5), 110.4 (C'-6), 123.4 (C'-7), 55.9 (OMe), 60.0 (OMe).

p-Bromobenzoate 15 from 14.—Compound 14 (15 mg) was acylated with p-bromobenzoyl chloride and dry pyridine at room temperature overnight, then extracted with CHCl<sub>3</sub> and purified by preparative tlc using C<sub>6</sub>H<sub>6</sub>-MeCN (9:1) as eluent to give the bromobenzoate 15 (16 mg): mp 173–175° [lit. (23) 175– 179°]; <sup>1</sup>H nmr  $\delta$  7.87 (2H, d, J = 8.5 Hz, H'-4 and H'-6), 7.56 (2H, d, J = 8.5 Hz, H'-3 and H'-7), 5.73 (1H, d, J = 10.5 Hz, H-6), 5.38 (1H, br d, J = 10.5 Hz, H-5), 2.92 (2H, m, H-1 and H-3), 1.82 (1H, br s, H-15), 1.39 (3H, s, H-14), 1.08 (3H, s, H-12), 0.91 (3H, s, H-13).

Compound **16**.—Colorless oil:  $[M]^+$  at m/z 402.2415,  $C_{24}H_{34}O_5$  requires 402.2406; ir  $\nu$  max 3015, 3000, 2960, 2920, 2860, 1690, 1590, 1505, 1455, 1410, 1280, 1260, 1170, 1015, 940, 930 cm<sup>-1</sup>; <sup>1</sup>H nmr  $\delta$  7.68 (1H, dd, J = 1.9, 8.4 Hz, H'-7), 6.90 (1H, d, J = 8.4 Hz, H'-6), 5.70 (1H, d, J = 10.5 Hz, H-6), 5.56 (1H, d, J = 1.8 Hz, H'-3), 5.44 (1H, br d, J = 10.5 Hz, H-5), 3.94 (6H, s, 2-OMe), 3.14 (1H, t, J = 13 Hz, H-3), 2.98 (1H, dd, J = 3.1, 11 Hz, H-1), 1.77 (3H, br s, H-15), 1.32 (3H, s, H-14), 1.14 (3H, s, H-12), 0.94 (3H, s, H-13); ms at m/z (rel. int.) [M]<sup>+</sup> 402 (3), 359 (1), 290 (1), 237 (2), 220 (14), 205 (6), 192 (4), 182 (100); <sup>13</sup>C nmr see Table 1.

*Himachalol* [17].—Mp 63–65° [lit. (18) 67–68°]; [M]<sup>+</sup> at m/z 222.2003,  $C_{15}H_{26}O_1$  requires 222.2003; ir  $\nu \max 3590$ , 3440, 3000, 2920, 2815, 1530, 1510, 1500, 1465, 1435, 1430, 1385, 1375, 1365, 1235, 1060, 1025, 908, 860 cm<sup>-1</sup>; <sup>1</sup>H nmr  $\delta$  5.48 (1H, d, J = 5.8 Hz, H-2), 1.58 (3H, br s, H-3), 1.15 (3H, s, H-15), 0.89 (3H, s, H-12), 0.76 (3H, s, H-13); <sup>13</sup>C nmr  $\delta$  133.3 (s), 125.7 (d), 76.1 (s), 51.9 (d), 43.6 (d), 41.3 (t), 38.4 (s), 36.4 (t), 33.5 (q), 32.6 (q), 31.7 (t), 26.8 (q), 23.7 (q), 22.3 (t), 19.9 (t); ms m/z (rel. int.) [M]<sup>+</sup> 222 (1), [M – Me]<sup>+</sup> 207 (4), [M – H<sub>2</sub>O]<sup>+</sup> 204 (36), 189 (12).

Compounds 21 and 23.—A mixture of alcohols 20 and 22 (40 mg) was treated overnight with  $Ac_2O/P$  pyridine at room temperature, and after workup, the reaction mixture was purified by cc on Si gel using  $C_6H_6$ -MeCN (19:1) as eluent to give a diacetate 21:  $[M]^+$  at m/z 280.0952,  $C_{14}H_{16}O_6$  requires 280.0952; <sup>1</sup>H nmr  $\delta$  6.69 (1H, d, J = 1.5 Hz, H-Ar), 6.58 (1H, d, J = 1.5 Hz, H-Ar), 5.96 (2H, s, OCH<sub>2</sub>O), 5.51 (1H, t, J = 6.8 Hz, H-COAc), 2.27 (3H, s, ArOAc), 2.03 (3H, s, OAc), 1.76 (2H, m, -CH<sub>2</sub>-), 0.84 (3H, t, J = 7.4 Hz, Me); ms m/z (rel. int.)  $[M]^+$  280 (34),  $[M - C_2H_2O]^+$  238 (46), 221 (17), 209 (19), 196 (60), 179 (28); and a monoacetate 23: <sup>1</sup>H nmr  $\delta$  6.52 (1H, s, H-Ar), 6.49 (1H, s, H-Ar), 5.94 (2H, s, OCH<sub>2</sub>O), 5.53 (1H, t, J = 7 Hz, H-COAc), 3.89 (3H, s, OMe), 2.06 (3H, s, OAc), 1.96 (2H, m, -CH<sub>2</sub>-), 0.86 (3H, t, J = 7.5 Hz, Me); ms m/z (rel. int.)  $[M]^+$  252 (42), 223 (6), 210 (46), 195 (4), 193 (37), 181 (100).

*Hydrolysis of* **23**.—Compound **23** (15 mg) dissolved in MeOH (1.5 ml) was treated with a 3% solution of KOH in MeOH (2 ml) for 8 h. After usual extraction, tlc purification with  $C_6H_6$ -EtOAc (4:2) gave alcohol **22** (4 mg).

Reduction of 19.—Compound 19 (400 mg) in MeOH (10 ml) was treated with NaBH<sub>4</sub> (100 mg) at 0° for 2 h. Usual workup afforded the alcohol 22 (300 mg).

Compound 24.—Mp 95–97° (hexane/Me<sub>2</sub>CO);  $[M]^+$  at m/z 224.0687,  $C_{11}H_{12}O_5$  requires 224.0684; ir  $\nu$  max 1650, 1605, 1504, 1420, 1330, 1280, 1150, 1060, 1010 cm<sup>-1</sup>; <sup>1</sup>H nmr & 12.50 (1H, s, HOAr), 6.90 (1H, s, H-Ar), 6.10 (2H, s, OCH<sub>2</sub>O), 3.88 (3H, s, OMe), 2.95 (2H, q, J=7.5 Hz, -CH<sub>2</sub>-), 1.22 (3H, t, J = 7.5 Hz, Me); ms m/z (rel. int.)  $[M]^+$  224 (43),  $[M - Me]^+$  209 (1),  $[M - H_2O]^+$  206 (1), 195 (100), 180 (3), 167 (6).

Compound **25**.—Colorless oil (9 mg); <sup>1</sup>H nmr (80 MHz, CDCl<sub>3</sub>)  $\delta$  7.15 (2H, d, J = 8 Hz, H-Ar), 6.79 (2H, d, J = 8 Hz, H-Ar), 4.40 (1H, t, J = 5.5 Hz, O-CH-Ar), 3.60 (2H, d, J = 5 Hz, -CH<sub>2</sub>-O), 3.42 (2H, q, J = 7 Hz, O-CH<sub>2</sub>-), 1.84 (3H, t, J = 7 Hz, Me).

Acetylation of **25**.—Compound **25** (9 mg) dissolved in pyridine (1 ml) was treated with Ac<sub>2</sub>O (0.5 ml) at room temperature for 12 h. Extraction as usual gave the acetate **26** (8 mg): <sup>1</sup>H nmr,  $\delta$  7.27 (2H, d, J = 8.5 Hz, H-Ar), 6.99 (2H, d, J = 8.5 Hz, H-Ar), 4.44 (1H, t, J = 6 Hz, OAc-CH-Ar), 4.06 (2H, d, J = 6 Hz, -CH<sub>2</sub>-O), 3.34 (2H, q, J = 7 Hz, O-CH<sub>2</sub>-), 2.20 (3H, s, ArOAc), 1.97 (3H, s, OAc), 1.10 (3H, t, J = 7 Hz, Me); <sup>13</sup>C nmr  $\delta$  170.9 (s), 169.4 (s), 150.5 (s), 136.4 (s), 127.9 × 2 (d), 121.7 × 2 (d), 79.1 (d), 67.9 (t), 64.7 (t), 21.1 (q), 20.9 (q), 15.2 (q); ms m/z (rel. int.) [M - CH<sub>2</sub>Me]<sup>+</sup> 235 (1), [M - OCH<sub>2</sub>Me]<sup>+</sup> 221 (1), 193 (44), 179 (2), 175 (4), 152 (10), 151 (100).

Compound **27**.—Mp 105–107° (hexane);  $[M]^+$  at m/z 268.1096,  $C_{17}H_{16}O_3$  requires 268.1099; ir  $\nu$  max 1620, 1600, 1450, 1430, 1350, 1295, 1270, 1235, 1160, 1090, 1060, 820 cm<sup>-1</sup>; ms m/z (rel. int.)  $[M]^+$  268 (100),  $[M - Me]^+$  253 (30),  $[M - 2Me]^+$  238 (10), 225 (57), 210 (69), 195 (20), 165 (42), 139 (52).

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